

# Easy, gentle and controlled ligand immobilization using Click Coupling

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## Immobilization made easy with Click Chemistry

Probably anyone who has ever performed an SPR experiment was initially faced with the problem of efficient and reproducible ligand immobilization.

The starting point is always the question of which immobilization chemistry is most compatible with the ligand, how to achieve optimal and reproducible ligand densities for a given application, and how to preserve the ligand's activity. In the end, it's all about which immobilization strategy generates the highest data quality in your specific SPR experiment.

In this newsletter, we introduce Click Chemistry as a versatile and easy-to-use coupling strategy and point out why Click Coupling is more than just another covalent immobilization method.

## Drawbacks of the state of the art

Covalent immobilization via EDC/NHS - the traditional workhorse of ligand immobilization on biosensor chips - has several negative attributes: First, the necessity for a buffer compatible with electrostatic pre-concentration, which is non-physiological, usually with low pH and low ionic strength. Second, the difficulty in controlling the immobilization level, which is hard to forecast in the first experiment and impossible to adjust later on. In addition, there are less obvious factors, such as random coupling sites and cross-linking of the ligand, which may adversely affect the ligand's binding characteristics.

Besides EDC/NHS coupling, affinity based ligand immobilization via streptavidin-biotin linkage has become one of the most popular immobilization methods for SPR biosensing: Though this approach provides some compelling features including convenient handling and excellent immobilization control in physiological conditions, the negative effects of a bulky linker like streptavidin should never be underestimated. Pre-immobilized streptavidin occupies a significant portion of the available space within the evanescent field, resulting in a reduced ligand immobilization yield and entailing the risk of hindered analyte diffusion.

**[A complete list of advantages and disadvantages you can find here.](#)**

	Covalent EDC/NHS	Covalent click-chemistry	Streptavidin/ biotin	Ni-NTA/ His6
Non-specific binding	Low	Very low	Moderate	High
Baseline drift after immobilization	No	No	Low	Moderate – High**
Targeting a specific immobilization level	Difficult	Easy	Easy	Moderate
Repeated immobilization cycles	No	Yes	Yes	Yes
Pre-concentration mandatory for immobilization	Yes	No	No	No
No crosslinking during immobilization	No	Yes*	Yes*	Yes
Quenching of reactive groups after immobilization required	Yes	No	No	No
Buffer restrictions during immobilization and measurement	Yes	No	No	Yes
Excluded volume by functional moiety	Low	Low	Very high	Low

\*If the conjugation ratio of the protein with the respective label is < 1.

\*\*NiHC moderate, NiD / NTA high.

## The challenge

Is there an immobilization chemistry which combines the simplicity of biotin–streptavidin immobilization, the immobilization yields of EDC/NHS chemistry, and is in addition highly selective & reproducible, sterically undemanding, and easy to apply?

## The solution

For about 20 years, Click Chemistry has gained increasing popularity among chemists, due to its outstanding reaction kinetics, excellent selectivity and superior stability in physiological environments. To make this unique bioconjugation technology available for the demanding ligand immobilization in SPR biosensing, XanTec's experienced surface chemists modified polycarboxylate sensor coatings with a very small, bioinert azido ( $N_3$ ) group, while the ligand was labeled with a low molecular weight cyclooctyne linker (DBCO), a procedure comparable to ligand biotinylation. Via a simple alkyne azide cycloaddition, these groups can react in a quantitative and highly selective manner, forming a stable covalent triazole linkage.

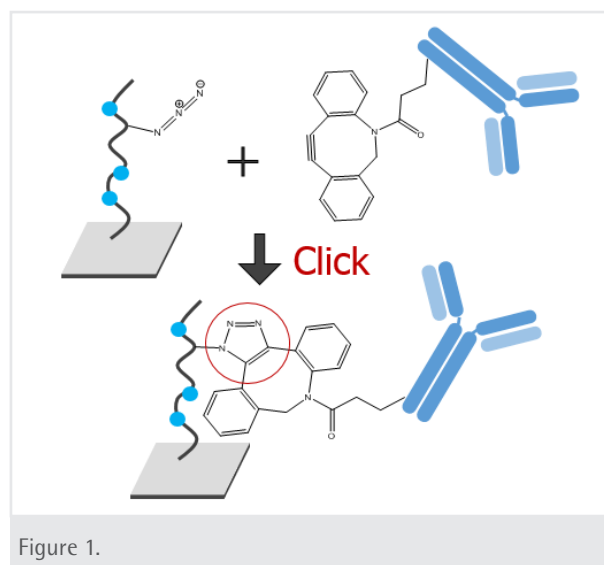


Figure 1.

Let's see how your SPR experiments can benefit from Click Chemistry:

## Advantages of Click Coupling

### Maximum convenience:

- The sensor chip comes pre-activated, ready-to-use, and can be stored for years with practically no loss of activity over time.
- Labeling of the protein is straightforward: Add linker to the protein, incubate for a few hours, desalt, and start immobilization.
- Stored frozen, the linker is reactive for many months.

### Maximum data quality:

- Very low non-specific binding.
- Higher sensitivity and lower diffusion artefacts compared with streptavidin-modified sensor chips due to small linker size [Fig. 2].

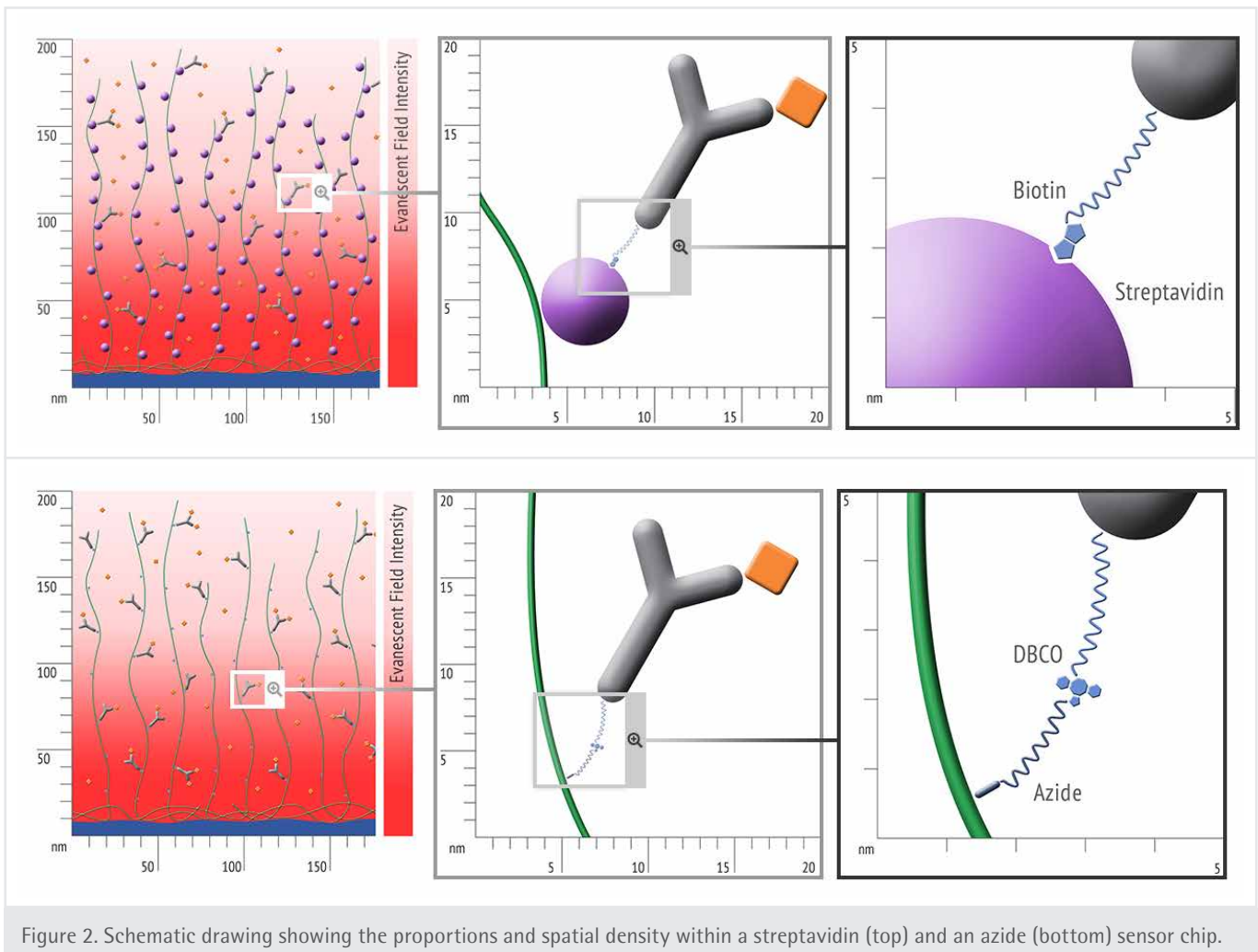


Figure 2. Schematic drawing showing the proportions and spatial density within a streptavidin (top) and an azide (bottom) sensor chip.

#### Maximum activity:

- One linker per ligand molecule maximizes ligand activity and avoids crosslinking.
- Labeling and immobilization of the ligand can be performed in physiological conditions.

#### Maximum control:

- The immobilization level can be adjusted at any time – even after the first interaction cycles.
- High stability and selective reactivity – no need to quench remaining reactive groups.

#### Maximum flexibility:

- Use classic electrostatic preconcentration to maximize immobilization yields.
- Use physiological immobilization conditions if ligand stability is critical.

## You might ask, “My SPR experiments have always worked, why should I use Click Coupling?”

Many of the disadvantages of other immobilization methods are not obvious without using sophisticated data analysis techniques, even though they negatively affect the result of an experiment and the quality of the information obtained. More obvious artifacts, such as high background, non-stoichiometric interaction due to a high fraction of inactive ligand, or difficulty in controlling immobilization levels are tolerated and taken as given, although the user is well aware that the data quality could be better.

### Who should use Click Coupling and why:

- Anyone who is still using **EDC/NHS chemistry** – to avoid crosslinking, (partial) deactivation of the ligand, and to increase the reproducibility of the immobilization level.
- Anyone who uses biotinylated ligands with **streptavidin** sensor chips – to avoid diffusion artefacts, to minimize non-specific binding, and to increase the immobilization yield.
- Anyone who needs **reproducible** immobilization levels for standardized assays to increase precision and reproducibility in QC/QA.
- Anyone whose protein tolerates only certain buffers to remain **stable and active**. No buffer restrictions during immobilization and analysis.
- Anyone who is struggling with **high non-specific binding** on NTA sensor chips.
- Anyone who intends to **increase the immobilization level** or add a second ligand later on, even within the actual binding experiment. Using Click Coupling, immobilization levels that are too low are a thing of the past.

## Protocol for DBCO labeling

The DBCO-PEG<sub>4</sub>-NHS ester introduces dibenzylcyclooctyne (DBCO) functionality to available amine groups (in lysine residues) on a protein. The PEG<sub>4</sub> spacer enhances solubility in water as well as in commonly used organic solvents like DMSO or DMF. Moreover, the spacer enhances the accessibility of the amine site, which may be buried within the protein.

As with every NHS ester, the DBCO-PEG<sub>4</sub>-NHS ester is moisture sensitive and hydrolyzes readily in aqueous buffers. Therefore, we recommend preparing a stock solution in dry solvent immediately before use; that is usually stable for several days. Do not use buffers containing primary amines like Tris, glycine or azides, as they will either react with the NHS ester or with the DBCO.

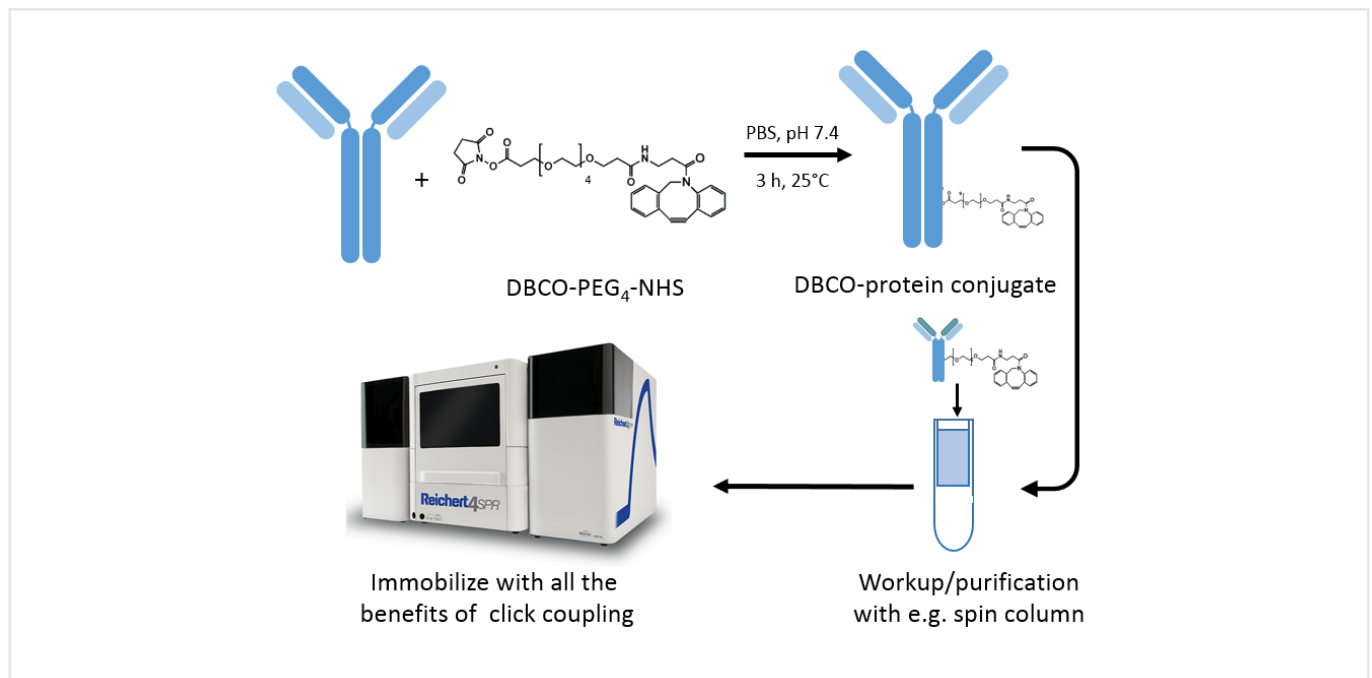
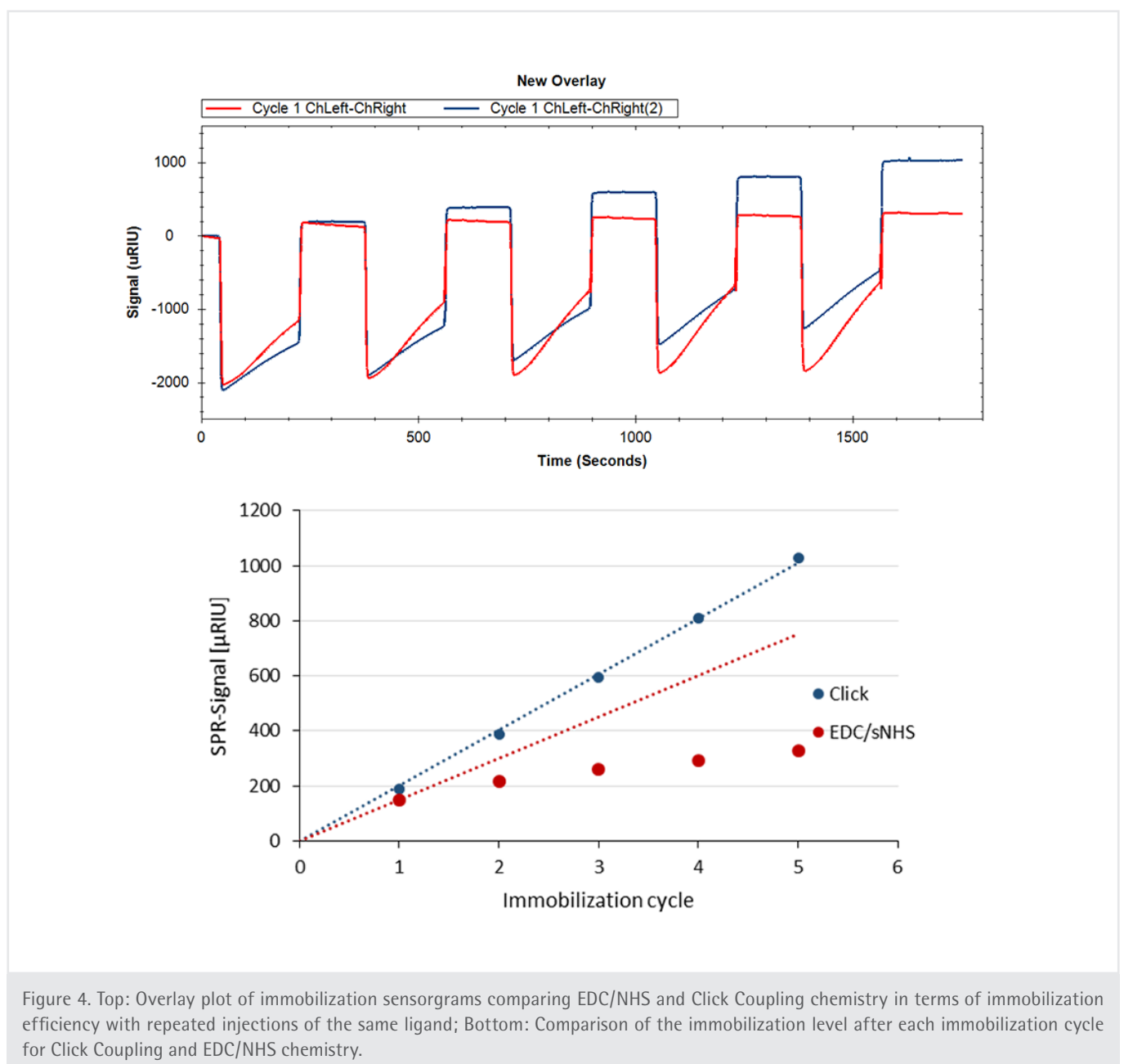


Figure 3. Overview of steps for ligand preparation before immobilization with Click Coupling.

- Prepare protein solution in phosphate-buffered saline or another suitable buffer (Fig. 3).
- Immediately before use, prepare the DBCO-PEG<sub>4</sub>-NHS reagent at 10 mM in anhydrous DMSO or DMF.
- For labeling of the protein, start with a 3-fold molar excess of the labeling reagent. This prevents most proteins from being over-labeled, which causes crosslinking on the sensor chip. For lysine-poor proteins, a higher molar excess of the labeling reagent might be necessary, and vice versa for lysine-rich proteins.
- Incubate the reaction for 3 h at room temperature or 8 h on ice.
- Remove non-reacted reagent by desalting with a spin column or dialysis.

## Immobilization of DBCO-labeled protein

Insert an azido-sensor chip according to instructions of the instrument manufacturer. There are no restrictions on the composition of the running buffer, however added azide should be avoided during the immobilization steps. There is no need for an activation step for the azido-surface. Simply inject the DBCO-labeled protein at concentrations in the range of 10–200  $\mu\text{g/mL}$ . Reaching a targeted immobilization level will require longer contact times for proteins in physiological conditions compared with electrostatic pre-concentration conditions. Please keep in mind that you can always add more ligand later, as long as sufficient unreacted azido-groups are present (Fig. 4). A quenching step is not necessary, as the unreacted azido functionalities on the sensor chip are non-reactive and bioinert.



## Non-specific binding (NSB)/interaction analysis/stability

### Non-specific binding

Data quality in interaction analysis is influenced by many extrinsic factors. Instrument status, purity and activity of the reactants, the quality of the sensor chip, and especially the ability to reduce non-specific binding to a minimum are important for obtaining high quality data. Comparing the NSB of surfaces for traditional EDC/NHS chemistry (CMD and HC) with those that are able to bind labeled/tagged biomolecules (streptavidin/Ni-NTA/azido) reveals significant differences (Fig. 5).

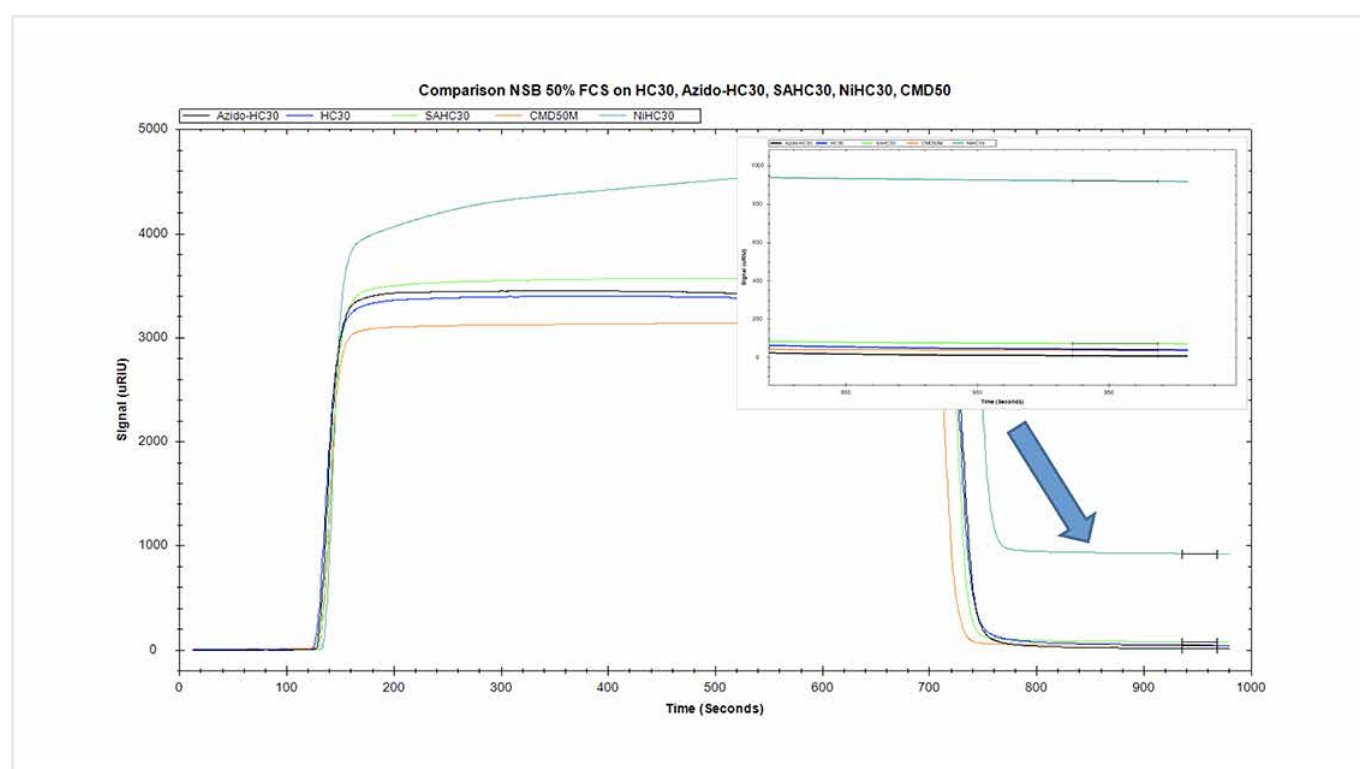


Figure 5. Overlaid and normalized SPR sensorgrams demonstrate the non-specific interaction of 50% fetal calf serum (FCS) in PBS on different surfaces. FCS was injected for 10 min at a flow rate of 20  $\mu\text{L}/\text{min}$ . Surfaces: CMD50M, HC30M, azido-HC30M, streptavidin-HC30M (SAHC30M), and NTA-HC30M (NiHC30M).

XanTec has tested the NSB of differently functionalized 30-nm linear polycarboxylate hydrogel chip coatings. In addition, a very similar sensor chip coated with 50 nm carboxymethyl dextran was tested. The azido-modified polycarboxylate surface has the lowest non-specific binding of all surfaces (8.3  $\mu\text{RIU}$ ). Even the unmodified carboxymethyl dextran (CMD, 37.5  $\mu\text{RIU}$ ) and linear polycarboxylate (HC, 42.1  $\mu\text{RIU}$ ) surfaces display a higher background. With a NSB of 72.6  $\mu\text{RIU}$ , the contribution of pre-immobilized streptavidin to NSB is clearly visible from comparison with the corresponding unmodified surface. The highest NSB was for the NTA-modified chip (920.2  $\mu\text{RIU}$ , NiHC30M).

Compared with the new azido derivative, non-specific binding on the streptavidin-modified surface is almost one order of magnitude higher, and on the Ni-NTA surface NSB is more than two orders of magnitude higher.

## Stability/storage

In contrast to NHS preactivated sensor chips, azido surfaces are very stable and can be stored frozen for at least several months. Also, the DBCO-linker is stable for months if stored frozen. Fig. 6 shows the immobilization yield of a batch of DBCO-conjugated protein A/G on azido-functionalized HC30M sensor chips (all chips from one batch).

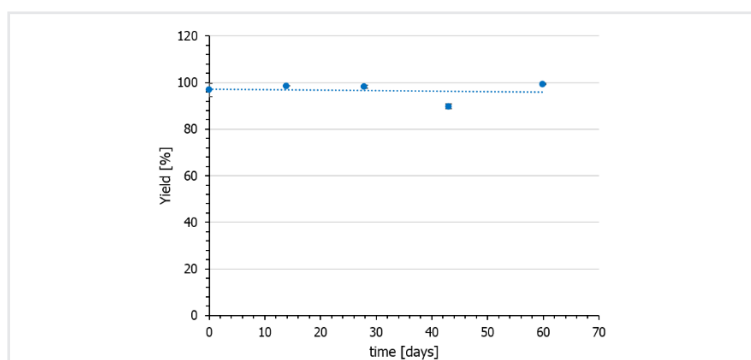


Figure 6. Immobilization yield of DBCO-labeled protein on azido-functionalized HC30M surface over the course of 2 months.

## Interaction analysis

For interaction analysis, a well-known reference assay for SPR instruments and surfaces was chosen. Carbonic anhydrase II (CAII) was immobilized in a Reichert 4SPR system with EDC and sulfo-NHS (a more reactive ester than NHS) and with the new Click Coupling immobilization method on a plain and on an azido-modified HC30M sensor chip, respectively. The analyte sulphiride ( $M_w = 341$  Da) in PBS was injected at six different concentrations, each in triplicate, at 25°C (Fig. 7).

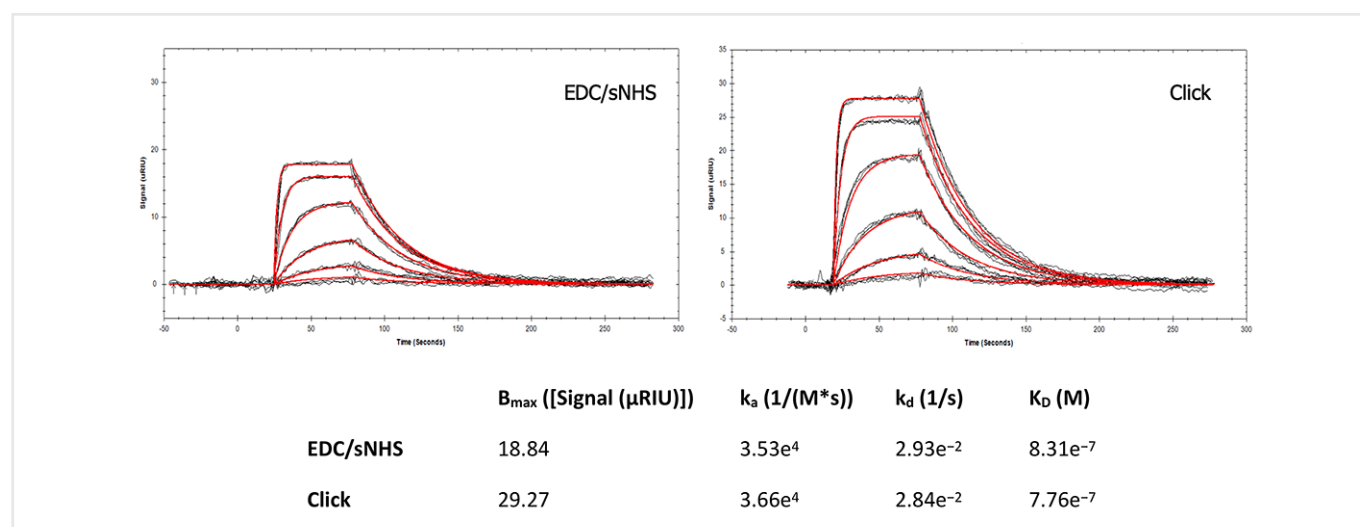


Figure 7. Sensorgrams of the interaction of immobilized CAII with sulphiride ( $M_w = 341$  Da) on a sensor chip with direct covalent immobilization of CAII via EDC/sNHS (left), and on an azido-sensor chip with DBCO-labeled CAII (right). The running buffer was PBS with 3% DMSO. Each analyte concentration was injected in triplicate. Measurements were performed at 25°C. Data shown are double referenced and DMSO corrected.

This side-by-side comparison shows that Click Coupling chemistry has no influence on the interaction of CAII and sulphiride compared with direct immobilization via EDC/NHS. All data are in good accordance with peer-reviewed, published data<sup>1,2</sup>. Slightly higher on- and off-rates for the Click-Coupling chemistry might indicate better accessibility of the ligand by the analyte due to the presence of the PEG<sub>4</sub> linker.



## Conclusion

- Click Coupling is an efficient, universal and robust immobilization chemistry. It is particularly advantageous for sensitive proteins requiring a specific buffer system (e.g. to maintain their activity), since labeling and immobilization have virtually no buffer restrictions.
- Also, this immobilization method is very well suited for standardized assays which need to be conducted using strictly identical protocols.
- The non-specific background is significantly lower with Click Coupling compared with other chemistries.
- Other common problems, such as multiple crosslinks, scouting for optimal immobilization conditions, and difficulties reaching a sufficient immobilization level are eliminated.
- The activity of the immobilized ligand increases significantly.
- Click-Coupling allows the user to increase the ligand density step-by-step and at any time – even after the first interaction cycles.

Click-coupling is a highly repeatable, robust and ligand-friendly immobilization chemistry, which significantly increases signal-to-noise ratio and data quality.

## Curious?

- **Get convinced of the advantages of Click Coupling and order Click chips today.**
- **As a complement to preactivated sensor chips, we offer a kit for convenient coupling and workup of ligands.**

## Surfaces for azide labelled ligands

Some expression systems are capable of site-specific integration of non-natural amino acids containing an azide moiety. For efficient, selective and site-directed Click Coupling of such ligands **XanTec** is also offering **DBCO modified sensor chips**.

## References

1. Navratilova, I., Papalia, G. A., Rich, R. L., Bedinger, D., Brophy, S., Condon, B., ... & Heutmekers, T. (2007). Thermodynamic benchmark study using Biacore technology. *Analytical biochemistry*, 364(1), 67-77.
2. Papalia, G. A., Leavitt, S., Bynum, M. A., Katsamba, P. S., Wilton, R., Qiu, H., ... & Giannetti, A. M. (2006). Comparative analysis of 10 small molecules binding to carbonic anhydrase II by different investigators using Biacore technology. *Analytical biochemistry*, 359(1), 94-105.

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# Protocol for DBCO-PEG<sub>4</sub>-NHS ester labeling

## Introduction

The DBCO-PEG<sub>4</sub>-NHS ester is an amine-reactive reagent which is soluble in organic solvents such as dimethylsulfoxide (DMSO) or dimethylformamide (DMF). Once dissolved in an organic solvent, the reagent is further diluted in a non-amine-containing aqueous buffer. DBCO-PEG<sub>4</sub>-NHS esters react efficiently with primary amino groups ( $-NH_2$ ) in neutral or slightly basic buffers to form stable amide bonds. Antibodies and proteins generally have multiple primary amines available as targets for labeling with NHS-activated DBCO-PEG reagents as they contain multiple lysine residues in addition to the N-terminus of each polypeptide. To avoid crosslinking of the hydrogel during immobilization, the labelling degree of the protein should be  $\leq 1$ .

## Product information

The DBCO-PEG<sub>4</sub>-NHS ester is moisture-sensitive (it is subject to hydrolysis of the NHS ester). Store the DBCO-PEG<sub>4</sub>-NHS ester-containing vials at  $-20^\circ\text{C}$ . Before use, equilibrate the vial to room temperature before opening/adding solvent to avoid moisture condensation onto the product.

Dissolve the DBCO-PEG<sub>4</sub>-NHS ester immediately before use in an anhydrous organic solvent. Discard any unused reconstituted reagent.

Avoid buffers containing primary amines (e.g., Tris, glycine or azides) as these compete with the intended reaction. If necessary, dialyze or otherwise desalt to exchange the protein sample into an amine-free buffer such as phosphate-buffered saline (PBS).

## Additional materials required

PBS: 0.1 M phosphate, 0.15 M sodium chloride, pH 7.2. Other non-amine-containing buffers at pH 7.0–8.0 may be used instead.

Quenching buffer: Tris-buffered saline (TBS): 1 M Tris, 0.15 M sodium chloride, pH 7.2. Glycine or other amine-containing buffers may be used instead.

Water-miscible anhydrous organic solvent such as DMSO or DMF

## Procedure for labeling protein with DBCO-PEG<sub>4</sub>-NHS ester

- Prepare protein in PBS or other non-amine-containing buffer
- Equilibrate the vial of DBCO-PEG<sub>4</sub>-NHS ester to room temperature before dissolving in organic solvent
- Dissolve 1–10 mg protein in 0.5–2 mL of PBS
- Immediately before use, prepare a 10 mM solution of DBCO-PEG<sub>4</sub>-NHS ester by adding 20 µL DMSO or DMF to the vial (0.1 mg)
- For labeling of the protein, start with a threefold molar excess of the labeling reagent. This prevents most proteins from being over-labeled, which causes crosslinking on the sensor chip. For lysine-poor proteins, a higher molar excess of the labeling reagent might be necessary, and vice versa
- Incubate the reaction at for 3 h at room temperature or 8 h on ice
- Remove the unreacted DBCO-PEG<sub>4</sub>-NHS ester by spin column or dialysis
- Store the DBCO-PEG labeled protein in the conditions that are optimal for the non-labeled protein

## Troubleshooting

- No reaction of labeled protein with azido-modified sensor chip:
  - Protein is not labeled
  - Possible reason: Hydrolysis of NHS ester
    - Allow the DBCO-PEG<sub>4</sub>-NHS ester-containing vial to equilibrate to room temperature before use
    - Prepare a new solution with anhydrous DMSO or DMF
    - Make sure that the buffers used are free from Tris, glycine, azides, and other primary amines

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